

**XE (F): Q. 1 – Q. 9 carry one mark each & Q. 10 – Q. 22 carry two marks each.**

- Q.1 Which one of the following polymers occurs naturally?  
 (A) Bakelite (B) Teflon (C) Cellulose (D) Perspex
- Q.2 The order of average molecular weights of a polymer is  
 (A)  $M_z > M_w > M_v > M_n$   
 (B)  $M_w > M_z > M_n > M_v$   
 (C)  $M_n > M_w > M_v > M_z$   
 (D)  $M_z > M_v > M_n > M_w$
- Q.3 Rubbers are a class of polymer known for  
 (A) High intermolecular forces  
 (B) High  $T_g$  polymers  
 (C) Crystalline polymers  
 (D) Low intermolecular forces
- Q.4 Nylon 6 is manufactured from  
 (A) Sebacic acid and hexamethylene diamine  
 (B) Caprolactam  
 (C) Adipic acid and hexamethylene diamine  
 (D) Caprolactone
- Q.5 Storage modulus and  $\tan \delta$  of a polymer are experimentally measured by  
 (A) Differential scanning calorimetry  
 (B) Thermogravimetric analysis  
 (C) Thermomechanical analysis  
 (D) Dynamic mechanical thermal analysis
- Q.6 A plastic bucket is manufactured by  
 (A) Compression moulding  
 (B) Injection moulding  
 (C) Extrusion  
 (D) Blow moulding
- Q.7 The monomers, A and B with reactivity ratios  $r_A$  and  $r_B$ , form alternate copolymers when,  
 (A)  $r_A = r_B = 0$  (B)  $r_A = r_B = 1$  (C)  $r_A > 1, r_B > 1$  (D)  $r_A < 1, r_B < 1$

Q.8 The degree of polymerization of a poly(methyl methacrylate) sample having number average molecular weight of 1,50,000 g/mol is \_\_\_\_\_.  
(C = 12, H = 1, O = 16 g/mol).

Q.9 If the heat of fusion of 100 % crystalline polyethylene is 290 mJ/mg, a sample of polyethylene with heat of fusion of 141 mJ/mg will have \_\_\_\_\_ % crystallinity.

Q.10 Match the following:

P. Butyl rubber	1. Metallocene polymerization
Q. Cold SBR	2. Cationic polymerization
R. Poly(ethylene terephthalate)	3. Redox polymerization
S. Polypropylene	4. Condensation polymerization

(A) P-3; Q-1; R-2; S-1  
(C) P-4; Q-3; R-1; S-2

(B) P-2; Q-3; R-1; S-4  
(D) P-2; Q-3; R-4; S-1

Q.11 Match the following:

P. Polyaramid	1. Baby-feeding nipple
Q. Polytetrafluoroethylene	2. Optical glasses
R. Polycarbonate	3. Non-stick cookware
S. Poly(dimethyl siloxane)	4. Bullet-proof jacket

(A) P-4; Q-3; R-2; S-1  
(C) P-4; Q-1; R-2; S-3

(B) P-2; Q-3; R-4; S-1  
(D) P-3; Q-4; R-2; S-1

Q.12 Flexible PVC tubes are used for watering. If some organic solvents are passed through this tube, it becomes stiff. This is due to the fact that the organic solvents

(A) plasticize PVC and raise  $T_g$ .  
(B) remove plasticizer and raise  $T_g$ .  
(C) remove plasticizer and lower  $T_g$ .  
(D) react with PVC and increase  $T_g$ .

Q.13 Match the following:

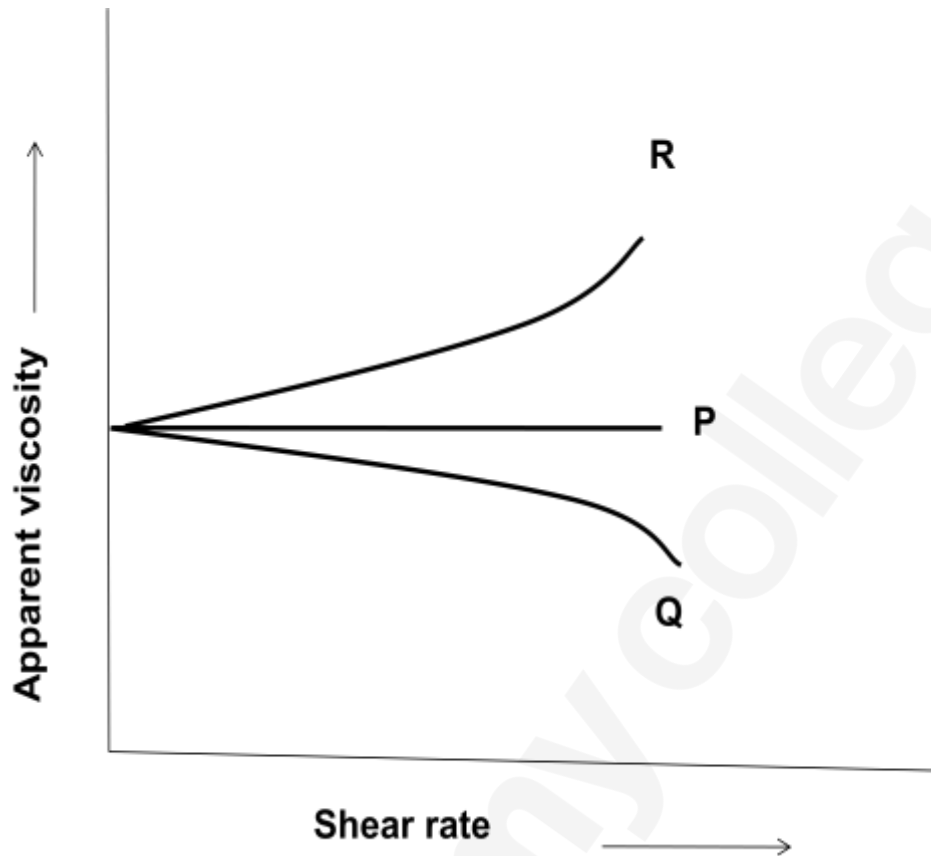
P. Plastic egg container	1. Injection moulding
Q. Water tank	2. Extrusion
R. Chair	3. Rotational moulding
S. Cable	4. Thermoforming

(A) P-3; Q-1; R-4; S-2  
(C) P-2; Q-3; R-4; S-1

(B) P-4; Q-3; R-2; S-1  
(D) P-4; Q-3; R-1; S-2



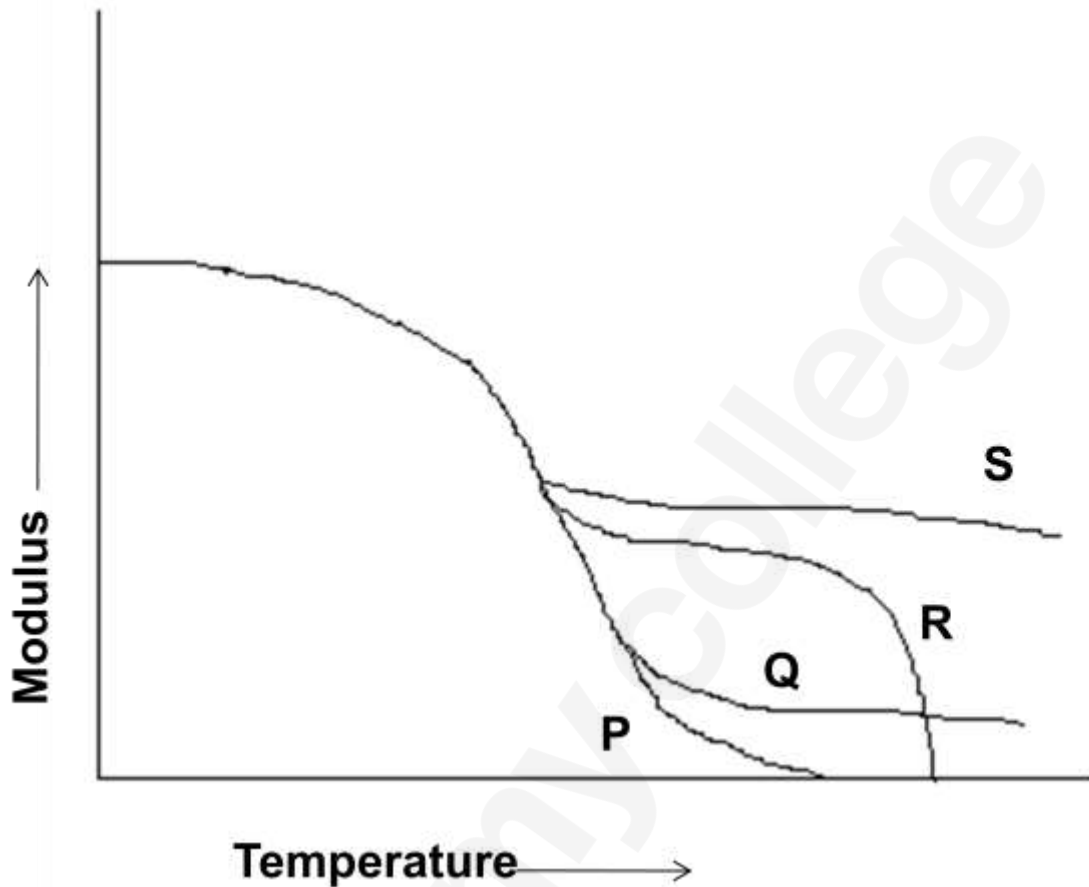
Q.16 The plot shows apparent viscosity versus shear rate of Newtonian, Dilatent and Pseudoplastic fluids. Based on this plot and the fluid behaviour as mentioned below, find out the correct combination.



1 = Dilatent fluid; 2 = Newtonian fluid, 3 = Pseudoplastic fluid

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|-------------------|-------------------|
| (A) P-1; Q-2; R-3 | (B) P-3; Q-1; R-2 |
| (C) P-2; Q-3; R-1 | (D) P-2; Q-3; Q-1 |

Q.17 Plot of the modulus versus temperature of different types of polymers is given below. Based on this plot and the nature of the polymers as mentioned below, find out the correct combination.



- 1 = An amorphous polymer of high molecular weight having entanglements
- 2 = An amorphous polymer of moderate molecular weight
- 3 = Highly crosslinked polymer
- 4 = Semi-crystalline polymer

- (A) P-2; Q-1; R-3; S-4
- (B) P-1; Q-2; R-3; S-4
- (C) P-2; Q-1; R-4; S-3
- (D) P-1; Q-3; R-4; S-2

Q.18 The  $T_g$  of homopolymers of A and B are  $+100\text{ }^\circ\text{C}$  and  $-70\text{ }^\circ\text{C}$  respectively. The  $T_g$  of a random copolymer of A and B having 40 wt% A and 60 wt% B is \_\_\_\_\_ $^\circ\text{C}$ .

Q.19 The number average molecular weight of a polymer prepared from  $\text{HO}(\text{CH}_2)_{14}\text{COOH}$  is 24,000 g/mol. The conversion of the monomer required to reach the above molecular weight is \_\_\_\_\_%. (C = 12, H = 1, O = 16 g/mol).

- Q.20 Glass fibers in nylon provide reinforcement. The modulus of elasticity for each component of the composite is;  $E_{\text{glass}} = 10.5 \times 10^6$  psi;  $E_{\text{nylon}} = 0.4 \times 10^6$  psi. If the nylon contains 30 vol % E-glass, the fraction of the applied force is carried by the glass fiber is \_\_\_\_\_. (Assume that both glass fiber and nylon have equal strain).
- Q.21 The solubility parameter of a polymer having cohesive energy density ( $E_{\text{coh}}$ ) 43870 J/mol and molar volume (V) 136 cm<sup>3</sup>/mol is \_\_\_\_\_ (J/cm<sup>3</sup>)<sup>1/2</sup>.
- Q.22 The heat of polymerization of styrene is 20 Kcal/mol. Heat of 5 x10<sup>5</sup> Kcal will be released on polymerization of \_\_\_\_\_ Kg of styrene (C = 12 and H = 1 g/mol).

**END OF THE QUESTION PAPER**